COURSE TITLE: General Chemistry I
COURSE NUMBER: CHM 1045

COURSE DESCRIPTION (with prerequisites):
The courses CHM 1045-1046 are designed to fulfill requirements in general chemistry for the first year in science, premedical, and engineering curricula. Includes units and measurements, chemical calculations, thermochemistry, gases, liquids, solids, atomic structure, and bonding. Prerequisite: CHM 1030 (with a grade of C or better) or one credit in high school chemistry and eligibility for MAC 1140 or a more advanced course. CHM 1045L should be taken concurrently. 3 semester hours credit.

NAME(S) OF INSTRUCTORS:
Jocelyn Wahlgren

EFFECTIVE ACADEMIC YEAR:
2011-12

REQUIRED TEXTBOOKS AND INSTRUCTIONAL SUPPLIES:
Can be purchased at the book store or online.

Or

GRADING POLICY:
The standing of a student in each course is expressed by one of the following letters and corresponding grading system:
A – 90 – 100
B – 80 – 89
C – 70 – 79
D – 60 – 69
F – 59 or less

The Chipola Catalog provides policies and procedures regarding the grading system. A student’s Grade Point Average is derived from the grading system/quality point scale.

ATTENDANCE AND WITHDRAWAL POLICIES:
Chipola College expects regular attendance of all students. Students who are absent from classes for any reason other than official college activities must satisfy the
instructor concerned that the absence was due to illness or other clearly unavoidable reasons. Otherwise, the student may suffer grade loss at the discretion of the instructor. Chipola policy allows each instructor to specify in the Instructor First Day Handout whether or not an absence is excusable and what affect the absence or tardy may have on the grade.

A student is allowed to repeat a course a maximum of three (3) times. On the third attempt a student (1) must bear the full cost of instruction (unless waived by Student Services), (2) cannot withdraw, and (3) must receive a grade.

MAKE-UP POLICY:
Chipola allows each instructor to specify in the Instructor First Day Handout the makeup policy.

ACADEMIC HONOR CODE POLICY:
Students are expected to uphold the Academic Honor Code. Chipola College’s Honor Code is based on the premise that each student has the responsibility to (1) uphold the highest standards of academic honesty in his/her own work; (2) refuse to tolerate academic dishonesty in the college community; and (3) foster a high sense of honor and social responsibility on the part of students. Further information regarding the Academic Honor Code may be found in the Chipola Catalog, Student Governance section.

STUDENTS WITH DISABILITIES POLICY:
Chipola College is committed to making all programs and facilities accessible to anyone with a disability. Chipola’s goal is for students to obtain maximum benefit from their educational experience and to effectively transition into the college environment. Students with disabilities are requested to voluntarily contact the Office of Students with Disabilities to complete the intake process and determine their eligibility for reasonable accommodations.

LIBRARY AND ON-LINE REFERENCE MATERIALS:
The library is a comprehensive learning resource center providing information in print, electronic, and multimedia format to support the educational objectives of the College. On-line catalogs, e-books and electronic databases can be accessed by using the LINCCWeb icon on the Chipola Library website at www.chipola.edu/library. If you have questions about database usage consult the “How to Use the Chipola Databases” on the Library website or call the Library at 850/718-2274 during regular hours. Library hours are posted each semester at the building entrance and on the Library website. See your Instructor First Day Handout for individual instructor recommendations and resources.

TECHNOLOGY RESOURCES:
The Information Technology Center, located in the library, is equipped with computer workstations. Lab hours are posted each semester at the building entrance and on the Library website. The ACE Lab, located in Building L, is available for tutoring and is equipped with computer workstations. Lab hours are posted each semester at the room
entrance. The college’s learning management system is **Desire 2 Learn** (d2l). Classes become available on d2l on the first day of the semester. It is the student’s responsibility to log onto the d2l system the first day of class to establish the first day of attendance and to check announcements. For further information, contact your instructor or the Director of Online Learning.

**ELECTRONIC DEVICE USAGE:**
All electronic devices such as cell phones, beepers, pagers, and related devices are to be silenced prior to entering classrooms and/or laboratories to avoid disruption. Should it become necessary for a student to leave his/her “device” on to send or receive an emergency call and/or text message, the student must inform the instructor prior to class. If the student finds it necessary to send and/or receive an emergency call and/or text message during class/lab time, he/she is instructed to take all books and belongings and step outside the classroom to deal with the situation. To minimize classroom disruption and the distraction to classmates, the student will not be permitted to reenter the classroom during that class period. Any time a test is being administered, all such devices must be turned off and put away. If a device is seen or heard during an exam, a score of zero will be given for that exam. Initial and repeated infractions may result in disciplinary action.

**DISCIPLINE SPECIFIC COMPETENCIES / LEARNING OUTCOMES:**
Explore the History, Nature, Methods, and Limits of Science

NS-1 Use methods of scientific inquiry.
NS-2 Apply scientific principles.
NS-3 Identify differences among scientific ideas related to the history or nature of science.
NS-4 Examine issues and problems facing modern science, such as ethics, values, and public policies.
NS-5 Identify relationships between science and technology.

**LINKING COURSE-LEVEL STUDENT LEARNING OUTCOMES WITH DISCIPLINE-SPECIFIC COMPETENCIES, ASSESSMENT METHODS, AND ARTIFACTS**

<table>
<thead>
<tr>
<th>COURSE-LEVEL STUDENT LEARNING OUTCOMES FOR CHM 1045*</th>
<th>DISCIPLINE-SPECIFIC GENERAL EDUCATION COMPETENCIES</th>
<th>ASSESSMENT METHODS FOR COURSE LEVEL STUDENT LEARNING OUTCOMES</th>
<th>ARTIFACTS FOR AA PROGRAM ASSESSMENT</th>
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</table>
| • Make use of the language and terminology of chemistry  
• Demonstrate a working knowledge of the fundamental | NS-1  
NS-1, NS-2, NS-3, NS-4, | T, CF, H  
T, CF, H, Exp. | No artifact will be submitted for program assessment as |

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3
concepts in chemistry
- Differentiate between competing theories that explain chemical concepts such as the valence bond and molecular orbital theories of bonding
- Solve problems relating to chemical concepts
- Interpret data from chemical experiments graphically or numerically
- Identify how technology is used in chemical processes

| NS-5 | T, CF, H |
| NS-2, NS-3 | T, CF, H, Exp. |
| NS-1, NS-2, NS-3, NS-4, NS-5, M-1, M-2, M-3, M-4, M-5 | T, CF, H, Exp. |
| NS-1, NS-2, NS-3, NS-5, M-1, M-2, M-3, M-4, M-5 | T, CF, H, Exp. |
| NS-1, NS-4, NS-5 | T, CF, H, Exp. |

**Assessment Codes**
- T = Tests
- Pre/Post = Pre- and Post-Tests
- OT = Objective Tests
- UT = Unit Tests
- Q = Quizzes
- F = Final Examination
- CF = Cumulative Final
- EX = Departmental Exam
- SE = Nat’l or State Standardized Exam
- RPT = Report/Presentation
- SP = Skills Performance
- SD = Skills Demonstration
- W = Writing Assignments
- E = Essays
- DE = Documented Essays
- RP = Research papers
- J = Jury
- R = Recital
- Proj. = Projects
- Exp. = Experiments
- Cap. Proj. = Capstone Project
- Cap. Course = Capstone Course
- Prac. = Practicum
- Intern. = Internship
- H = Homework
- PS = Problem Solving
- DB = Discussion Board
- BO = Behavioral Observation
- Clin. = Clinicals
- CS = Case Study
- CP = Case Plan
- Port. = Portfolio
- Obs. = Teacher Observation
- Sk. Check = Skills Check-off
- JP = Judged
- Curriculum Frameworks

**Means of Accomplishing Student Learning Outcomes:**
Students will participate in a lecture course utilizing, power point, board work, group work, and a classroom response system. Students will complete online homework using WebAssign. Students will be asked to study in groups outside of class time.

**Assignment and/or Course Outline**
*Additional learning outcomes in this course include the following*
Florida Teacher Certification Exam Competencies in General Science 5-9: 1.x, 2.4, 2.5, 3.1, 3.3, 3.5, 3.6, 3.9, 3.10, 3.11, 9.1, 9.3, 9.6
Florida Teacher Certification Exam Competencies in Biology 6-12: 1.8, 1.10, 1.11, 1.13, 3.3
Next Generation Natural Science Sunshine State Standards: 5N13, 5N15, 5N21, 5N22, 5P81, 5P82, 5P83, 5P84, 5P10.1, 5P10.3, 5P11.2, 6N22, 6N31, 6N32, 6N33, 7N21, 7N32, 7P11.3, 7P11.4, 8N13, 8P81, 8P82, 8P83, 8P84, 8P85, 8P86, 8P87, 8P88, 8P89, 8P91, 8P92, 912E58, 912N11, 912N34, 912P81, 912P82, 912P83, 912P84, 912P85, 912P87, 912P88, 912P89, 912P8.11, 912P10.1, 912P10.2, 912P10.5, 912P10.6, 912P10.7, 912P10.9, 912P10.18, 912P12.10
The topics covered in this course will include: the metric system, significant figures, conversion factors, atomic theory, covalent and ionic compounds, nomenclature of inorganic compounds, stoichiometry, replacement reactions, redox reactions, acid-base reactions, solution stoichiometry, gases, thermochemistry, atomic structure, periodicity, Lewis dot structures, VSEPR, hybridization, and molecular orbital theory.

See your Instructor First Day Handout for individual instructor assignment schedule.